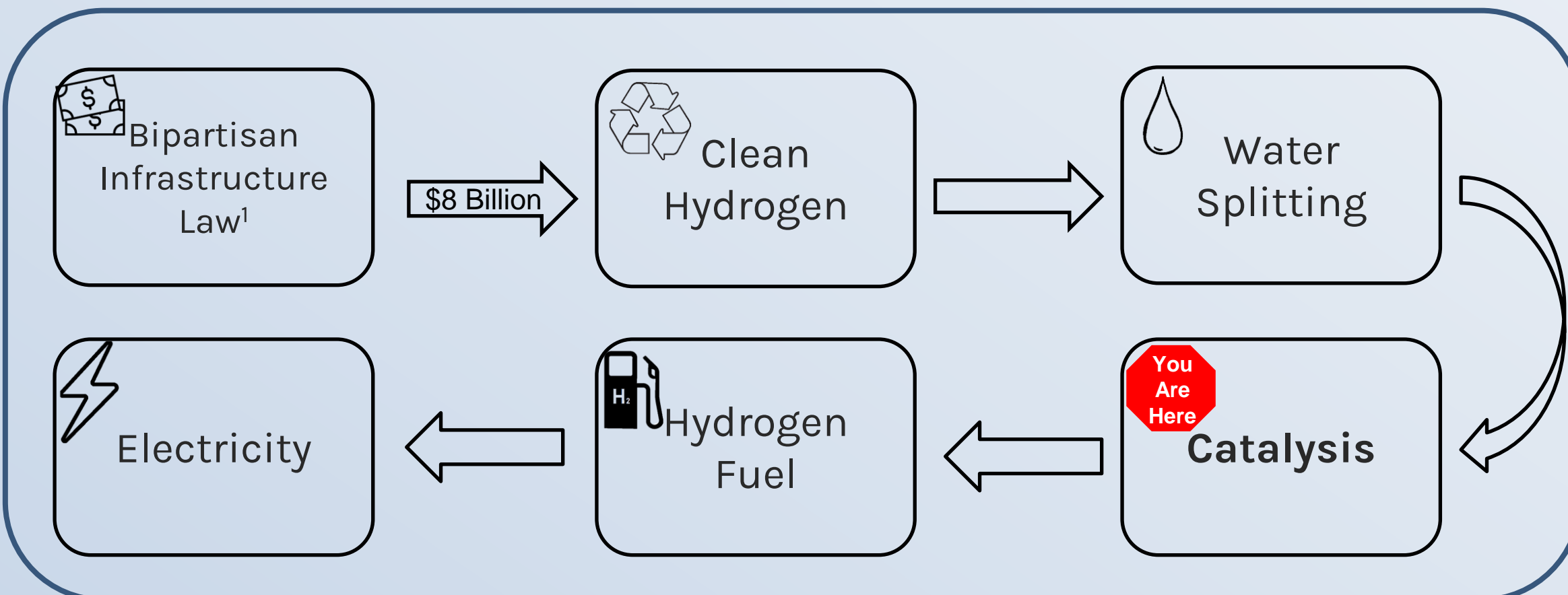


Engineering nanoscale catalysts for clean hydrogen fuel production

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I. The Big Picture

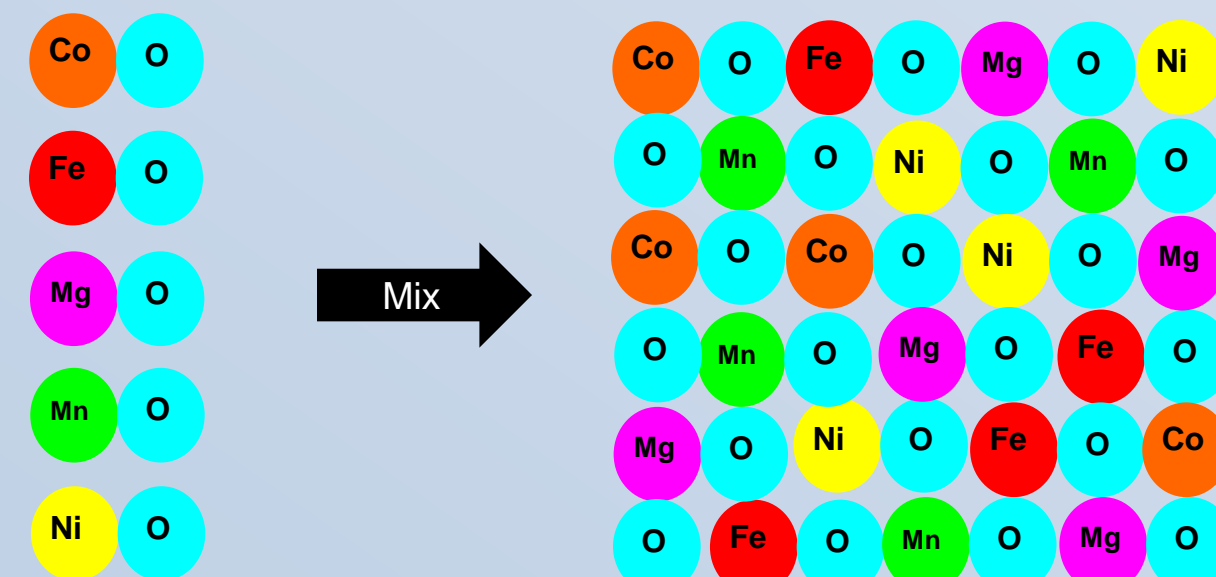


II. Background

- Water Splitting: Takes water and makes it into hydrogen and oxygen $H_2O \rightarrow H_2 + O_2$
- This process is inefficient. To increase the efficiency, a catalyst is used
- A catalyst increases the speed of a chemical reaction
- Common catalysts for water splitting include ruthenium oxide and iridium oxide (RuO_2 and IrO_2)²
- Ru and Ir are both expensive, rare, and precious metals
- Per oz they cost \$475 and \$4,800 respectively, the price of gold is \$1,860³
- Cheap and earth abundant water splitting catalyst are needed**

III. What This Group is Doing

- High-entropy metal oxides (**HEOs**)
- HEOs are composed of multiple metal elements in about equal proportions
- They have shown promise as a catalysis for water splitting because of their unique complex surfaces⁴
- A rock-salt structure (**RSS**) is a type of crystal structure
- We have a lot experience manipulating different RSS
- Our Goal is to make HEOs from earth abundant metals (such as nickel, iron, and magnesium) that are in a RSS**



IV. Results So Far

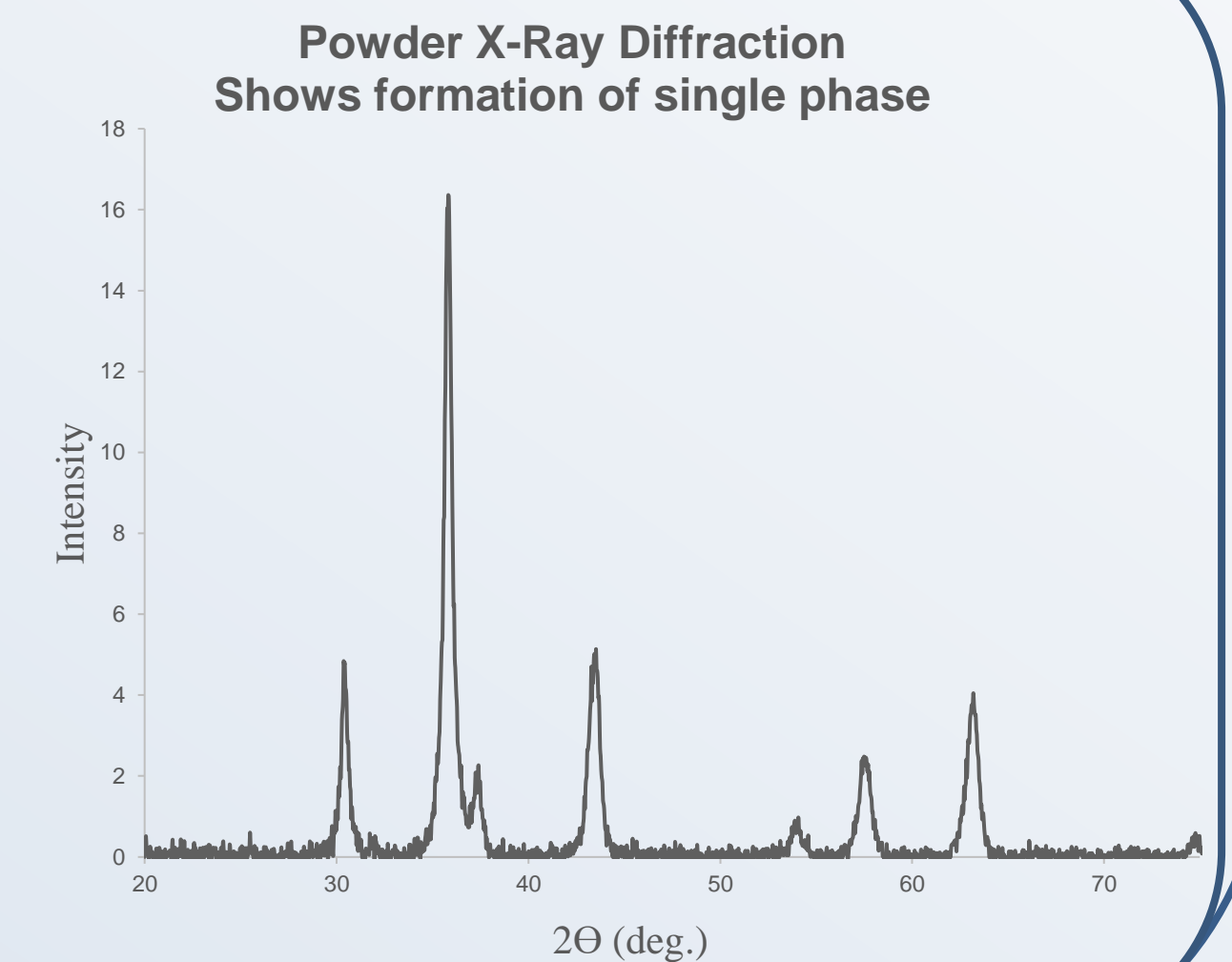
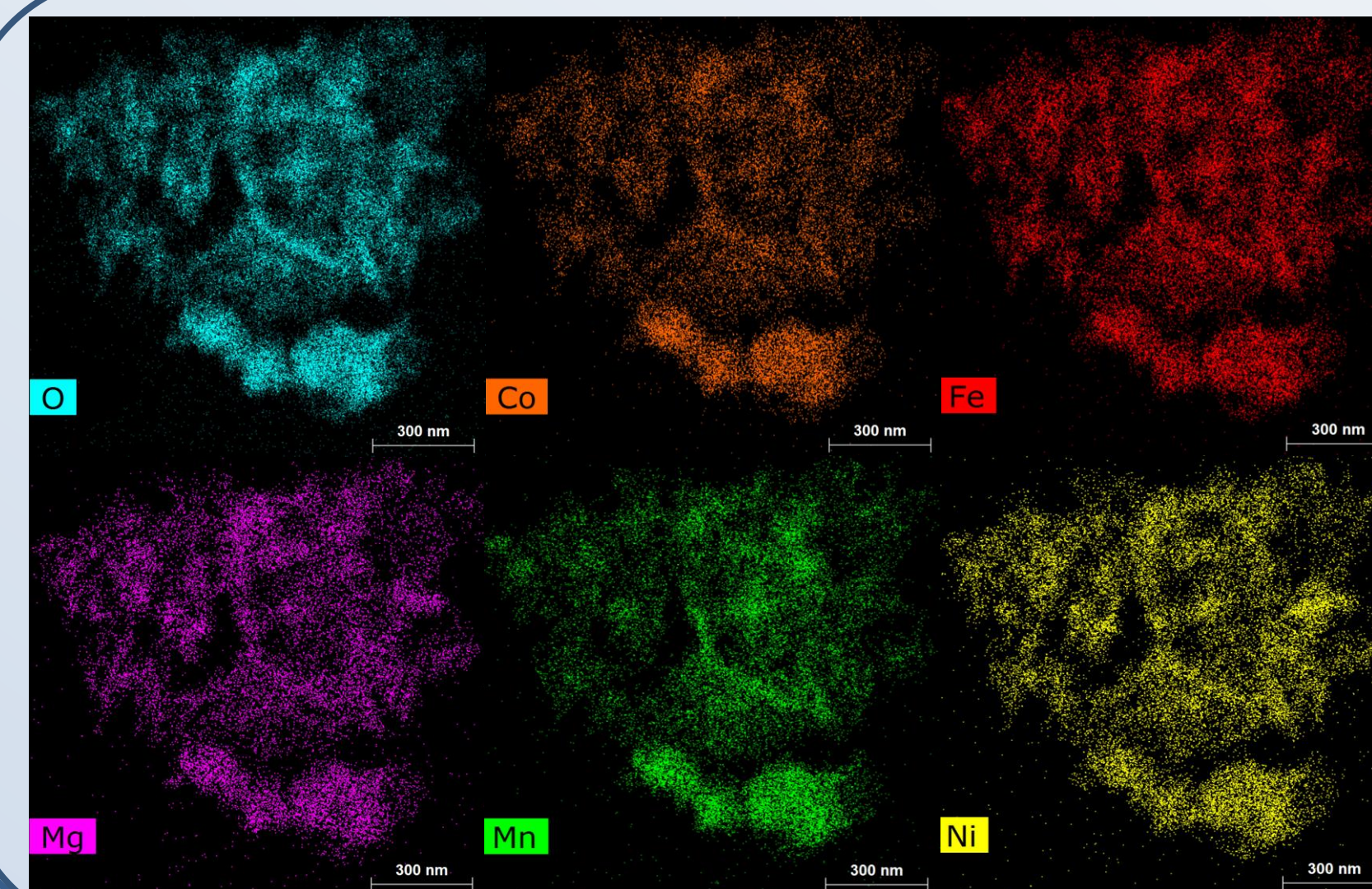
IV A. Overview

- We have made a HEO
- Just not with a RSS
- Our Current Work is trying to make an HEO that has a RSS. This will allow us to manipulate the HEO and make it an even better catalysis**

IV B. Details

- The micrograph (left) suggest the elemental distribution of a HEO
- The graph (right) confirms a HEO distribution

IV C. Data



V. Acknowledgements

- The Ryan Richards Research Group
- The Colorado School of Mines
- The ARCS Foundation
- Project Bridge, PDA, and SACNAS



VI. References

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